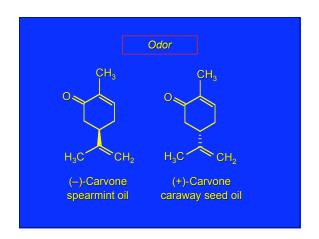


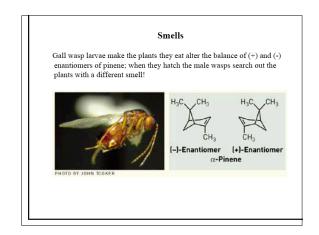
Physical Properties of Enantiomers

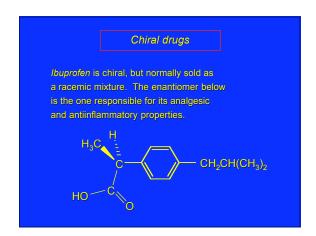
Properties of enantiomers

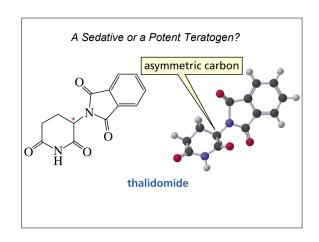
Physical properties are the same:
melting point, boiling point, density, etc

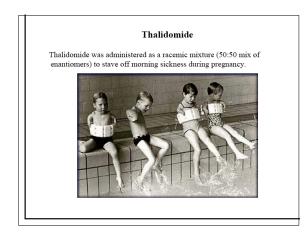
Some others are different:
properties that depend on the shape of molecule eg. biological-physiological and optical properties.

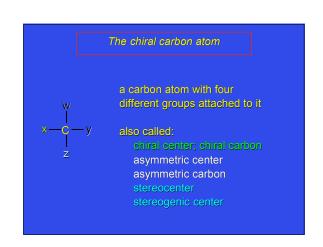












Chirality and chiral carbons

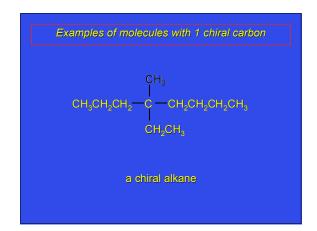
A molecule with a single stereogenic center is chiral.

2-Butanol is an example.

H

CH<sub>3</sub>—C—CH<sub>2</sub>CH<sub>3</sub>

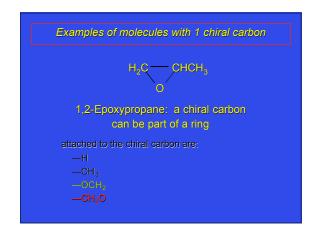
OH



Examples of molecules with 1 chiral carbon

OH

Linalool, a naturally occurring chiral alcohol

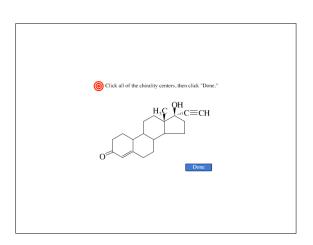


Limonene: a chiral carbon

CH<sub>3</sub> carbon can be part of a ring

attached to the chiral carbon are:

-H
-C=CH<sub>2</sub>
-CH<sub>2</sub>CH<sub>2</sub>
-CH<sub>3</sub>CH=
-C=



## A molecule with a single chiral carbon must be chiral. But, a molecule with two or more chiral carbons may be chiral or it may not. Ie'll return to this when we consider molecules with more than one chiral carbon atom.